

## Primary cycloadducts of 1,10-phenanthrolinium and phthalazinium phenacylides with DMAD

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**Abstract**—Primary cycloadducts *cis-*2 and *trans-*6 of monosubstituted cycloimmonium phenacylides 1 and 5 with DMAD have been obtained for the first time. They isomerise stereospecifically and regiospecifically by prototropic rearrangement to dihydro derivatives *cis-*3 and *trans-*7, respectively. The new heterocyclic system of pyrrolo[1,2-a][1,10]phenanthroline was illustrated by a series of derivatives (8a−f). The ethyl (8b) and isopropyl (8c) esters exhibit helical chirality by ¹H NMR. © 2001 Elsevier Science Ltd. All rights reserved.

The unstable monosubstituted heteroaromatic *N*-ylides obtained in situ by deprotonation of the corresponding cycloimmonium salts in the presence of base (mostly triethylamine) are 1,3-dipoles which undergo cycloaddition with acetylenic dipolarophiles resulting in the formation of a fused five-membered nitrogen heterocycle. Aromatisation by loss of two hydrogen atoms occurs spontaneously during the reaction in the presence of oxygen. As a matter of fact, the primary cycloaddition product, a 2,5-dihydropyrrolo derivative has never been isolated or detected in the reaction mixture.<sup>1-3</sup> A claim<sup>4</sup> to have obtained such a product from an unsubtituted phthalazinium methylide lacks sufficient information (NMR spectral characterisation by only one chemical shift).

We have succeeded in obtaining the primary cycload-ducts cis-2 and trans-6 from the title ylides 1 and 5 with dimethyl acetylenedicarboxylate (DMAD) by replacing the usual aprotic solvents used in the cycloaddition with methanol. The nucleophilicity of triethylamine is reduced by solvation so that further isomerisation by a contact ion-pair mechanism is obstructed. The reactions were performed with ice cooling and yields over 90% were registered. Under the same conditions, the quino-

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linium and isoquinolinium phenacylides led directly to the corresponding dehydrogenated pyrroloazines.

The cycloadditions are stereospecific with the cycloadducts *cis-2* and *trans-6* corresponding to the ylide geometries *syn-1* (W-dipole) and *anti-5* (S-dipole), respectively. NMR spectra of the free ylides could not be registered because of their instability.

Protrotropic rearrangements of cis-2 and trans-6 take place quantitatively to give the dihydro derivatives cis-3 and trans-7, respectively, at room temperature in the presence of triethylamine in deuterochloroform. Considering the reacting dihydropyrrolo moieties of cis-2 and trans-6 as cis-trans isomers, their allylic rearrangements to cis-3 and trans-7 are stereospecific and regiospecific at the same time. In the former case the 3-H atom, geminal to the benzoyl group, migrates while in the latter the angular hydrogen atom (3a-H) is shifting. No traces of equilibrium products could be detected by <sup>1</sup>H NMR. Each one of the isomers cis-2 and trans-6, under thermodynamic control, could afford two pairs of cis-trans regioisomers. In fact one single compound results from each primary cycloadduct. The pyrroline moieties of cis-3 and trans-7 are regioisomers to each other with opposite configurations. This possibility was anticipated by Epiotis<sup>5</sup> in 1978 who introduced a new term 'chorochemistry' which collectively describes stereochemistry and regiochemistry. We might say the reactions occur 'chorospecifically'. In the strict sense, trans-7 is 'choroisomeric'

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with *cis-***9a** which has also been identified (see below). The exclusive formation of one isomer is also a good diagnosis for the *anti* or *syn* conformation of the initial reacting ylide.

The primary cycloadduct *trans-6* melted at 157–160°C in a sealed tube and after resolidification the melting

point rose to 257–259°C on the microplate, which corresponded to the dehydrogenated product **8a**. Compound *trans*-**7** (mp 222–225°C) displayed the same behaviour. The *cis*-cycloadduct **2** (mp 163–164°C) on melting rearranged to *cis*-**3** (mp 129–131°C) which dehydrogenated to **4**<sup>6</sup> by refluxing in acetic acid.

Figure 1.

The relative stability of the dihydro derivatives *cis-***3** and *trans-***7** allowed us to draw the conclusion that they were the only by-products of the spontaneous dehydrogenation of *cis-***2** and *trans-***6**.

The reaction with phenanthroline was initially planned with the purpose of synthesising derivatives of a new heterocyclic system, pyrrolo[1,2-a][1,10]phenanthroline. The second nitrogen of the phenanthroline could not be made to react. Yields of over 80% were obtained for 8a-f<sup>7</sup> by carrying out the reaction in methylene chloride at room temperature. <sup>1</sup>H NMR measurement on the crude products indicated the presence of the dihydro derivative cis-9 in appreciable quantities (up to 80%). Its source is syn-5 due to the protic polar solvent methanol. Refluxing the mixture in ethanol for a short time led to the dehydrogenation product 8. However, in the case of quinolinium acetylmethylide an appreciable percentage (60%) of trans-10 was obtained which could be separated by chromatography unlike the other dihydro derivatives which aromatised on the column. Therefore, the reacting species of the initial ylide is *anti*. The formation of isomers of the type trans-7 and trans-10 by ylide cycloaddition has not been observed until now.

Structures were assigned on the basis of the following criteria: the coupling constants of the two protons of the pyrroline nucleus, <sup>1</sup>H and <sup>13</sup>C NMR chemical shifts of the corresponding CH and N-CH groups, of the carbonyl groups and also of the enaminic quaternary carbons.

The most characteristic feature of 6 is the unusually large trans-homoallylic coupling,  $J_{1,3a} = 7.3$  Hz. This has been already encountered in trans-2,5-dihydropyrrolines in the range 7.0-7.6 Hz.8-11 Its magnitude was explained by a dual path, the heteroatom being implicated in the mechanism of interaction by a synergistic four-bond coupling.<sup>8</sup> The *cis*-homoallylic coupling in 2 has a value of 1.5 Hz. The structures are confirmed by the presence of two N-CH groups, a benzoyl group bound to a saturated carbon, two esters groups grafted on a double bond and by the lack of a quaternary enaminic carbon. Supplementary evidence was given by COSY, HETCOR and NOE experiments (Fig. 1). 12 For isomers cis-3 and cis-9e the following coupling constants were measured  $J_{1,10b}=13.2$  and  $J_{3,3a}=13.8$  Hz. The vicinal and allylic coupling constants of protons in positions 3a, 4 and 5 of cis-9e are quite similar to those of the dihydro derivative trans-6. They are not observed in trans-7 whose chemical shifts and vicinal coupling constant,  $J_{1,2}=4.5$  Hz, have values close to those of the analogous *trans*-10 ( $J_{1,2}=4.2$  Hz). All these coupling constants have close values to those of the corresponding dihydroindolizines derivatives (4.0–7.0 and 13-14 Hz)<sup>13,14</sup> and also to those of the analogous 2-pyrrolines (5.0–7.0 and 13.0–13.4 Hz)<sup>9</sup> to which the configurations trans and cis have been assigned, respectively. Compounds 3, 7, 9 and 10 possess one single N-CH group and also an enaminic quaternary carbon, the high field values observed for the 3-C<sub>q</sub> in 7 and 10 being explained by the major contribution of the pyridinium canonical form 11. This also explains the higher  $\delta$  values for the protons of the N-CH groups. The 1-H protons of the pyrrolophenanthrolines 6 and 7 are strongly deshielded being in close vicinity to the nitrogen atom in position  $11.^{12}$ 

Several dihydro derivatives of pyrrolo[2,1-a]isoquinoline<sup>13,15</sup> and pyrrolo[1,2-b]pyridazine<sup>16,17</sup> are described in literature and were obtained as mixtures with the dehydrogenated products, all corresponding to the same type of structure cis-3 ( $J_{\rm vic}$ =13–14 Hz). Also erroneous structures were proposed. <sup>18,19</sup> They can be diagnosed now as originating from a syn-ylide.

The <sup>1</sup>H NMR spectra at room temperature of the diethyl (**8b**) and diisopropyl (**8c**) esters of the new heterocyclic system exhibited helical chirality. The energy barrier of **8b** was calculated from the coalescence temperature and was found to be  $\Delta G^{\ddagger}_{60^{\circ}\text{C}} = 16.5$  kcal/mol.

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- 70.1 (1-C); 89.9 (3- $C_q$ ); 165.6 (3-E); 172.6 (2-E); 201.8 (Ac).
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